organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Andreas F. G. Glawar,^a David J. Watkin,^a* Gangadhar J. Sanjayan,^b George E. Tranter,^c Alison A. Edwards^c and George W. J. Fleet^d

^aChemical Crystallography, Chemical Research Laboratory, University of Oxford, Oxford OX1 3TA, England, ^bDivision of Organic Synthesis, National Chemical Laboratory, Pune 411 008, India, ^cBiological Chemistry, Division of Biomedical Sciences, Imperial College, London SW7 2AZ, England, and ^dDepartment of Organic Chemistry, Chemical Research Laboratory, Mansfield Road, Oxford OX1 3TA, England

Correspondence e-mail: david.watkin@chem.ox.ac.uk

Key indicators

Single-crystal X-ray study T = 190 KMean $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ R factor = 0.050 wR factor = 0.096 Data-to-parameter ratio = 20.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved

Isopropyl 2,5-anhydro-3,4-di-O-tert-butyldiphenyl-silyl-L-ribonate

Determination of the crystal structure of the title compound, $C_{40}H_{50}O_5Si_2$, firmly established its relative configuration and hence that of some related tetrahydrofuran carboxylates. The material crystallizes with Z' = 2. Except for the chiral centres, the two independent molecules are related by a pseudo-centre of symmetry.

Received 15 July 2005 Accepted 25 July 2005 Online 27 July 2005

Comment

The reaction of methanol with lactones containing 2-Otrifluoromethanesulfonates (trifluoromethanesulfonates) in the presence of either acid (Wheatley et al., 1993) or base (Choi et al., 1992) provides a general synthesis of methyl tetrahydrofuran-2-carboxylates. Such materials have been exploited in the preparation of sugar amino acids (SAAs) for use as peptidomimetics (Chakraborty et al., 2004; Grotenberg et al., 2004; Smith et al., 2003). Many THF SAA scaffolds are predisposed to form secondary structures in short oligomers (Claridge et al., 2005; Long et al., 1999, 2002; Hungerford et al., 2000). There are only limited reports of γ -peptides based on cyclic templates (Curran et al., 1996; Crisma et al., 2001). In a programme directed towards the synthesis of γ -THF SAAs, it was found that reaction of the δ -lactone trifluoromethanesulfonate (1) (Stewart et al., 2002) with methanol in the presence of sodium carbonate gave a mixture of the THF carboxylates (2) and (3). In order to ensure the correct assignment of the stereochemistry at C-2 in the epimers, (3) was converted to the crystalline disilyl ether (4), the structure of which is reported in this paper (Fig. 1).



The structure of (4) contains two molecules in the asymmetric unit (Z' = 2). Except for the 1,4-anhydroribonate units (which are chiral and therefore cannot be related by an improper operator), the molecules are related by a pseudo-centre of symmetry at $(\frac{1}{2}, \frac{1}{5}, \frac{3}{4})$. The absolute configuration of the material was known unambiguously from the synthesis; the Flack (1983) parameter is in agreement with this assignment.

The structure consists of molecular layers (Fig. 2) lying parallel to the bc plane, and characterized by a hydrophilic and a hydrophobic surface. The hydrophobic surface of one layer faces the equivalent surface of the adjacent layer (Fig. 3).





Figure 2 The crystal structure projected on to the *bc* plane.





The crystal structure projected along the b axis, showing two hydrophobic faces opposing each other. By symmetry, pairs of hydrophilic faces also oppose each other.

The H atoms were all located in a difference map, but those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry (C-H = 0.93–98 Å) and displacement parameters [$U_{\rm iso}$ (H) = 1.2–1.5 $U_{\rm eq}$ (parent atom)], after which they were refined with riding constraints. The pseudo-centre of inversion did not lead to any refinement problems.

Data collection: *COLLECT* (Nonius, 2001); cell refinement: *DENZO/SCALEPACK*; data reduction: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

Financial support (to AAE) from the EPSRC and (to GJS) from DST, New Delhi, for a BOYSCAST Fellowship, is gratefully acknowledged.

Figure 1

The structure of one molecule of the title compound with displacement ellipsoids drawn at the 50% probability level. All H atoms, except for H31, H41, H61 and H71, have been omitted for clarity. The H atoms are drawn with an arbitary radius.

Experimental

Epimer (3) was converted to the corresponding disilyl ether by standard procedures (Sanjayan *et al.*, 2003) and was crystallized from ethyl acetate–hexane (1:4).

Crystal data

$C_{40}H_{50}O_5Si_2$
$M_r = 667.01$
Monoclinic, P2 ₁
a = 17.2952 (2) Å
b = 10.7468 (2) Å
c = 20.4914 (4) Å
$\beta = 100.7864 (5)^{\circ}$
V = 3741.40 (11) Å ³
Z = 4
Data collection
Nonius KappaCCD diffractometer
ω scans
Absorption correction: multi-scan
(DENZO/SCALEPACK; Otwi-
nowski & Minor, 1997)
T = 0.70 T = 0.97

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.050$ $wR(F^2) = 0.096$ S = 0.9717372 reflections 848 parameters H-atom parameters constrained

27685 measured reflections

 $D_x = 1.184 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 8312 reflections $\theta = 5-30^{\circ}$ $\mu = 0.14 \text{ mm}^{-1}$ T = 190 KPrism, colourless $0.40 \times 0.20 \times 0.20 \text{ mm}$

17372 independent reflections 17372 reflections with $I > -3\sigma(I)$ $R_{int} = 0.027$ $\theta_{max} = 30.0^{\circ}$ $h = -24 \rightarrow 24$ $k = -15 \rightarrow 15$ $l = -28 \rightarrow 28$

$$\begin{split} &w = 1/[\sigma^2(F^2) + (0.02P)^2 \\ &+ 2.21P] \\ &where \ P = [max(F_o^2,0) + 2F_c^2]/3 \\ (\Delta/\sigma)_{max} = 0.003 \\ \Delta\rho_{max} = 0.70 \ e \ \text{\AA}^{-3} \\ \Delta\rho_{min} = -0.47 \ e \ \text{\AA}^{-3} \\ &\text{Absolute structure: Flack (1983),} \\ 17372 \ \text{Friedel pairs} \\ &\text{Flack parameter: 0.05 (7)} \end{split}$$

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). J. Appl. Cryst. 36, 1487.
- Chakraborty, T. K., Srinivasi, P., Tapadar, S. & Mohan, B. K. (2004). J. Chem. Sci. 116, 187–207.
- Choi, S. S., Myerscough, P. M., Fairbanks, A. J., Skead, B. M., Bichard, C. J. F., Mantell, S. J., Fleet, G. W. J., Saunders, J. & Brown, D. (1992). J. Chem. Soc. Chem. Commun. pp. 1605–1607.
- Claridge, T. D. W., Long, D. D., Baker, C. M., Odell, B., Grant, G. H., Edwards, A. A., Tranter, G. E., Fleet., G. W. J. & Smith, M. D. (2005). J. Org. Chem. 70, 2082–2090.
- Crisma, M., Moretto, A., Toniolo, C., Kaczmarek, K. & Zabrocki, J. (2001). Macromolecules, 14, 5048–5052.
- Curran, T. P., Chandler, N. M., Kennedy, R. J. & Keaney, M. T. (1996). Tetrahedron Lett. 37, 1933–1936.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Grotenberg, G. M., Timmerj, M. S. M., Llamas-Saiz, A. L., Verdoes, M., van der Marel, G. A., van Raaij, M. J., Overkleeft, H. S. & Overhand, M. (2004). J. Am. Chem. Soc. 126, 3444–3446.

- Hungerford, N. L., Claridge, T. D. W., Watterson, M. P., Aplin, R. T., Moreno, A. & Fleet, G. W. J. (2000). J. Chem. Soc. Perkin Trans. 1, pp. 3666– 3679.
- Long, D. D., Hungerford, N. L., Smith, M. D., Brittain, D. E. A., Marquess, D. G., Claridge, T. D. W. & Fleet, G. W. J. (1999). *Tetrahedron Lett.* 40, 2195–2198.
- Long, D. D., Smith M. D., Martin, A., Wheatley, J. R., Watkin, D. G., Muller, M. & Fleet, G. W. J. (2002). J. Chem. Soc. Perkin Trans. 1, pp. 1982–1998.
- Nonius (2001). COLLECT. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sanjayan, G. J., Stewart, A. J., Hachisu, S., Gonzalez, R., Watterson., M. P. & Fleet, G. W. J. (2003). *Tetrahedron Lett.* 44, 5847–5852.
- Smith, M. D., Claridge, T. D. W., Sansom, M. P. & Fleet, G. W. J. (2003). Org. Biomol. Chem. 1, 3647–3655.
- Stewart, A. J., Evans, R. M., Weymouth-Wilson, A. C., Cowley, A. R., Watkin, D. J. & Fleet, G. W. J. (2002). *Tetrahedron Asymmetry*, 13, 2667–2672.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). CAMERON. Chemical Crystallography Laboratory, Oxford, England.
- Wheatley, J. R., Bichard, C. J. F., Mantell, S. J., Son, J. C., Hughes, D. J., Fleet, G. W. J. & Brown, D. (1993). J. Chem. Soc. Chem. Commun. pp. 1065– 1067.